

Tris[2-methoxy-6-(4-methylphenyliminio-methyl)phenolato- κ^2O,O']tris(thiocyanato- κN)neodymium(III)

Jia-Lu Liu, Jin-Bei Shen and Guo-Liang Zhao*

Zhejiang Key Laboratory for Reactive Chemistry on Solid Surfaces, Institute of Physical Chemistry, Zhejiang Normal University, Jinhua, Zhejiang 321004, People's Republic of China, and College of Chemistry and Life Science, Zhejiang Normal University, Jinhua 321004, Zhejiang, People's Republic of China
Correspondence e-mail: sky53@zjnu.cn

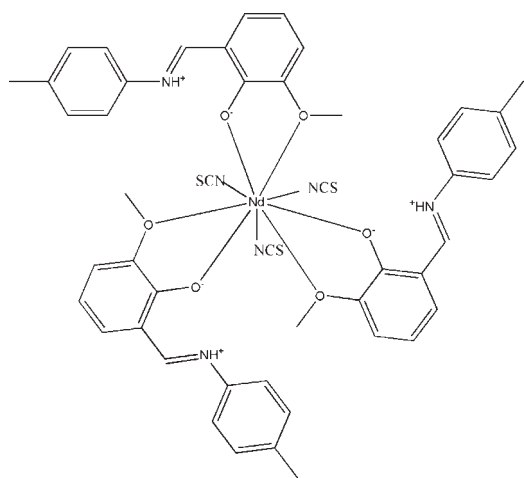
Received 1 November 2009; accepted 10 December 2009

Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(C-C) = 0.012$ Å; R factor = 0.060; wR factor = 0.190; data-to-parameter ratio = 20.6.

In the title compound, $[Nd(NCS)_3(C_{15}H_{15}NO_2)_3]$, the Nd^{III} ion is coordinated by three thiocyanate anions $[Nd-N = 2.489(8)–2.530(7)$ Å] and six O atoms $[Nd-O = 2.375(4)–2.843(5)$ Å] from three zwitterionic 2-methoxy-6-(4-methylphenyliminiomethyl)phenolate ligands in a tricapped trigonal-prismatic geometry. Intramolecular $N-H \cdots O$ hydrogen bonds occur. The crystal packing exhibits weak intermolecular $C-H \cdots S$ hydrogen bonds, $\pi-\pi$ interactions with a distance of $3.904(7)$ Å between the centroids of the aromatic rings, and voids of 101 Å³.

Related literature

For related structures, see: Wang & Chang (1994); Zhao *et al.* (2007); Xian *et al.* (2008); Li *et al.* (2008).



Experimental

Crystal data

$[Nd(NCS)_3(C_{15}H_{15}NO_2)_3]$
 $M_r = 1042.35$
Monoclinic, $P2_1/c$
 $a = 16.6107(3)$ Å
 $b = 14.2425(3)$ Å
 $c = 22.1582(4)$ Å
 $\beta = 105.972(1)^\circ$

$V = 5039.77(17)$ Å³
 $Z = 4$
Mo $K\alpha$ radiation
 $\mu = 1.21$ mm⁻¹
 $T = 296$ K
 $0.14 \times 0.08 \times 0.05$ mm

Data collection

Bruker APEXII diffractometer
44769 measured reflections
11604 independent reflections

5189 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.114$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.060$
 $wR(F^2) = 0.190$
 $S = 0.86$
11604 reflections

562 parameters
H-atom parameters constrained
 $\Delta\rho_{max} = 0.66$ e Å⁻³
 $\Delta\rho_{min} = -0.84$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N1-H1A \cdots O2$	0.86	1.86	2.566 (7)	138
$N2-H2A \cdots O4$	0.86	1.87	2.576 (6)	138
$N3-H3A \cdots O6$	0.86	1.89	2.589 (7)	137
$C38-H38A \cdots S1^i$	0.93	2.72	3.633 (6)	166

Symmetry code: (i) $-x + 3, -y + 1, -z + 3$.

Data collection: *APEX2* (Bruker, 2006); cell refinement: *SAINT* (Bruker, 2006); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV2653).

References

- Bruker (2006). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
Li, H.-Q., Xian, H.-D., Liu, J.-F. & Zhao, G.-L. (2008). *Acta Cryst. E* **64**, m1593–m1594.
Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.
Wang, G. & Chang, J.-C. (1994). *Synth. React. Inorg. Met. Org. Chem.* **24**, 623–627.
Xian, H.-D., Liu, J.-F., Li, H.-Q. & Zhao, G.-L. (2008). *Acta Cryst. E* **64**, m1422.
Zhao, G.-L., Shi, X. & Ng, S. W. (2007). *Acta Cryst. E* **63**, m267–m268.

supplementary materials

Acta Cryst. (2010). E66, m72 [doi:10.1107/S1600536809053124]

Tris[2-methoxy-6-(4-methylphenyliminomethyl)phenolato- κ^2O,O']tris(thiocyanato- κN)neodymium(III)

J.-L. Liu, J.-B. Shen and G.-L. Zhao

Comment

Schiff base ligands derived from substituted salicylaldehyde and their metal complexes have been widely investigated because of their structural features (Wang *et al.*, 1994). In this aspect we have synthesized several analogous Schiff bases transitional and rare metal complexes. In recent articles we have reported our research results (Zhao *et al.*, 2007; Xian *et al.*, 2008; Li *et al.*, 2008). Herein, we report the crystal structure of the title Nd³⁺ complex, (I).

In (I) (Fig. 1), the Nd^{III} is nine-coordinated by three nitrogen atoms from three thiocyanate ions and six O atoms from 2-(4-methylphenyliminomethyl)-6-methoxyphenolato (*L*). *L* ligands coordinate the Nd center with bidentate-chelate mode using oxygen atoms from deprotonated phenolic hydroxyl groups and methoxyl groups. The imino H atoms in *L* are involved in intramolecular N—H \cdots O hydrogen bonds (Table 1). The bonds between Nd^{III} and O atoms from phenolic hydroxyl groups are 2.375 (2)–2.393 (3) Å, which are shorter than those between Nd and O atoms of methoxyl groups (2.742 (3)–2.843 (3) Å). The Nd—N bond lengths are 2.489 (4)–2.530 (4) Å. The structure of (I) is similar to the structures of analogous complexes (Zhao *et al.*, 2007; Li *et al.*, 2008).

The crystal packing exhibits weak intermolecular C—H \cdots S hydrogen bonds (Table 1) and π – π interaction with the distance of 3.904 (7) Å between the centroids of aromatic rings.

Experimental

Reagents and solvents were of commercially available quality and were used without further purification. The Schiff base ligand was prepared in a high yield synthesis by condensation of *o*-vanillin and *p*-methylaniline and was recrystallized in ethanol before being used. 1 mmol Nd(NO₃)₃ (dissolved in methanol) was added dropwise into a methanol solution with 3 mmol 3-methoxy-*N*-salicylidene-*p*-toluidine under stirring and the mixture was continuously stirred at room temperature for 8 h to obtain a red solution. The deposit was filtered off and the solution was left standing for slow evaporation. Red crystals of the title compound were obtained after several days.

Refinement

All H atoms were positioned geometrically (C—H 0.93–0.96 Å, N—H 0.86 Å), and refined as riding, with $U_{\text{iso}}(\text{H}) = 1.2$ – $1.5U_{\text{eq}}(\text{C}, \text{N})$. The crystal packing exhibits voids of 101 Å³ centered at (0.19, 0.67, 0.24).

Figures

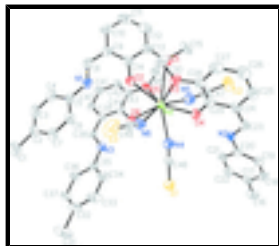


Fig. 1. The molecular structure of the title complex, showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level. H atoms omitted for clarity.

Tris[2-methoxy-6-(4-methylphenyliminomethyl)phenolato- κ^2O,O']tris(thiocyanato- κN)neodymium(III)

Crystal data

[Nd(NCS)₃(C₁₅H₁₅NO₂)₃]

$M_r = 1042.35$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 16.6107$ (3) Å

$b = 14.2425$ (3) Å

$c = 22.1582$ (4) Å

$\beta = 105.9720$ (10)°

$V = 5039.77$ (17) Å³

$Z = 4$

$F(000) = 2124$

$D_x = 1.374$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 2091 reflections

$\theta = 1.4$ – 27.6 °

$\mu = 1.21$ mm⁻¹

$T = 296$ K

Block, red

$0.14 \times 0.08 \times 0.05$ mm

Data collection

Bruker APEXII
diffractometer

Radiation source: fine-focus sealed tube
graphite

ω scans

44769 measured reflections

11604 independent reflections

5189 reflections with $I > 2\sigma(I)$

$R_{int} = 0.114$

$\theta_{max} = 27.6$ °, $\theta_{min} = 1.3$ °

$h = -21$ → 21

$k = -18$ → 18

$l = -28$ → 28

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.060$

$wR(F^2) = 0.190$

$S = 0.86$

11604 reflections

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.1P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{max} = 0.001$

562 parameters

$$\Delta\rho_{\max} = 0.66 \text{ e } \text{\AA}^{-3}$$

0 restraints

$$\Delta\rho_{\min} = -0.84 \text{ e } \text{\AA}^{-3}$$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Nd	1.21267 (2)	0.27661 (3)	1.393189 (17)	0.04942 (15)
C1	1.7474 (6)	0.2005 (8)	1.6278 (6)	0.132 (4)
H1B	1.7970	0.1935	1.6140	0.197*
H1C	1.7517	0.1616	1.6639	0.197*
H1D	1.7416	0.2649	1.6386	0.197*
C2	1.6709 (6)	0.1709 (6)	1.5751 (5)	0.090 (3)
C3	1.6775 (5)	0.1410 (7)	1.5180 (6)	0.098 (3)
H3B	1.7303	0.1368	1.5115	0.118*
C4	1.6086 (5)	0.1167 (6)	1.4699 (4)	0.079 (2)
H4A	1.6150	0.0969	1.4315	0.095*
C5	1.5300 (5)	0.1220 (5)	1.4793 (4)	0.064 (2)
C6	1.5214 (5)	0.1515 (5)	1.5365 (4)	0.066 (2)
H6A	1.4686	0.1552	1.5430	0.079*
C7	1.5906 (6)	0.1754 (6)	1.5834 (4)	0.082 (2)
H7A	1.5842	0.1951	1.6218	0.099*
C8	1.4479 (5)	0.0514 (5)	1.3821 (4)	0.065 (2)
H8A	1.4939	0.0182	1.3772	0.078*
C9	1.3712 (5)	0.0448 (5)	1.3347 (4)	0.069 (2)
C10	1.3636 (7)	-0.0112 (6)	1.2797 (5)	0.097 (3)
H10A	1.4097	-0.0448	1.2752	0.116*
C11	1.2900 (8)	-0.0158 (8)	1.2340 (5)	0.110 (4)
H11A	1.2860	-0.0531	1.1988	0.132*
C12	1.2212 (7)	0.0342 (7)	1.2392 (4)	0.096 (3)
H12A	1.1713	0.0307	1.2073	0.115*
C13	1.2253 (5)	0.0892 (5)	1.2909 (4)	0.066 (2)
C14	1.3004 (5)	0.0943 (5)	1.3399 (4)	0.0613 (19)
C15	1.0798 (5)	0.1339 (7)	1.2583 (4)	0.098 (3)
H15A	1.0791	0.0822	1.2304	0.147*
H15B	1.0642	0.1904	1.2344	0.147*
H15C	1.0408	0.1221	1.2823	0.147*

supplementary materials

C16	0.9337 (6)	0.8193 (6)	1.4168 (4)	0.092 (3)
H16A	0.9816	0.8487	1.4085	0.138*
H16B	0.9289	0.8393	1.4570	0.138*
H16C	0.8841	0.8368	1.3848	0.138*
C17	0.9441 (5)	0.7145 (5)	1.4168 (3)	0.0619 (19)
C18	0.8840 (5)	0.6555 (6)	1.4276 (4)	0.069 (2)
H18A	0.8360	0.6809	1.4348	0.083*
C19	0.8936 (4)	0.5584 (5)	1.4279 (3)	0.0601 (19)
H19A	0.8524	0.5195	1.4354	0.072*
C20	0.9633 (4)	0.5209 (5)	1.4171 (3)	0.0523 (17)
C21	1.0259 (4)	0.5778 (5)	1.4075 (3)	0.0604 (19)
H21A	1.0746	0.5520	1.4018	0.072*
C22	1.0147 (5)	0.6738 (5)	1.4067 (4)	0.071 (2)
H22A	1.0560	0.7124	1.3991	0.085*
C23	0.9246 (4)	0.3560 (5)	1.4198 (3)	0.0612 (19)
H23A	0.8711	0.3726	1.4217	0.073*
C24	0.9442 (4)	0.2599 (5)	1.4191 (3)	0.0576 (19)
C25	0.8829 (5)	0.1900 (6)	1.4185 (5)	0.089 (3)
H25A	0.8296	0.2076	1.4200	0.107*
C26	0.9016 (6)	0.0991 (7)	1.4158 (5)	0.105 (3)
H26A	0.8606	0.0539	1.4142	0.126*
C27	0.9813 (5)	0.0716 (6)	1.4151 (4)	0.083 (3)
H27A	0.9926	0.0079	1.4132	0.099*
C28	1.0438 (4)	0.1343 (5)	1.4174 (4)	0.0600 (19)
C29	1.0261 (4)	0.2302 (5)	1.4194 (3)	0.0495 (16)
C30	1.1531 (5)	0.0226 (5)	1.4252 (4)	0.086 (3)
H30A	1.1104	-0.0169	1.4333	0.129*
H30B	1.2027	0.0188	1.4598	0.129*
H30C	1.1656	0.0022	1.3875	0.129*
C31	1.6013 (7)	0.4363 (7)	1.7675 (4)	0.122 (4)
H31A	1.5565	0.4274	1.7866	0.183*
H31B	1.6238	0.4984	1.7765	0.183*
H31C	1.6446	0.3910	1.7840	0.183*
C32	1.5683 (6)	0.4236 (6)	1.6970 (4)	0.079 (2)
C33	1.4863 (6)	0.3989 (6)	1.6687 (4)	0.085 (3)
H33A	1.4496	0.3904	1.6933	0.102*
C34	1.4576 (5)	0.3865 (6)	1.6047 (4)	0.073 (2)
H34A	1.4018	0.3714	1.5862	0.088*
C35	1.5109 (4)	0.3963 (5)	1.5688 (3)	0.0514 (17)
C36	1.5935 (4)	0.4198 (5)	1.5956 (4)	0.063 (2)
H36A	1.6301	0.4266	1.5707	0.075*
C37	1.6216 (5)	0.4334 (6)	1.6600 (4)	0.075 (2)
H37A	1.6773	0.4493	1.6783	0.090*
C38	1.5238 (4)	0.3633 (4)	1.4624 (3)	0.0523 (17)
H38A	1.5819	0.3674	1.4765	0.063*
C39	1.4875 (4)	0.3426 (4)	1.3990 (3)	0.0482 (16)
C40	1.5387 (4)	0.3304 (5)	1.3584 (4)	0.064 (2)
H40A	1.5964	0.3373	1.3739	0.077*
C41	1.5055 (5)	0.3089 (6)	1.2974 (4)	0.072 (2)

H41A	1.5403	0.3005	1.2714	0.087*
C42	1.4185 (5)	0.2993 (5)	1.2729 (3)	0.0610 (19)
H42A	1.3953	0.2865	1.2305	0.073*
C43	1.3680 (4)	0.3090 (5)	1.3120 (3)	0.0522 (17)
C44	1.4000 (4)	0.3316 (5)	1.3758 (3)	0.0483 (16)
C45	1.2416 (5)	0.3000 (7)	1.2278 (4)	0.091 (3)
H45A	1.2821	0.3112	1.2050	0.137*
H45B	1.1993	0.3478	1.2178	0.137*
H45C	1.2162	0.2397	1.2166	0.137*
C46	1.2340 (4)	0.5040 (5)	1.4736 (3)	0.0591 (19)
C47	1.0439 (6)	0.3648 (7)	1.2736 (4)	0.090 (3)
C48	1.3000 (4)	0.1762 (5)	1.5462 (3)	0.0577 (18)
N1	1.4566 (4)	0.1018 (4)	1.4318 (3)	0.0628 (16)
H1A	1.4113	0.1262	1.4365	0.075*
N2	0.9766 (3)	0.4219 (4)	1.4181 (2)	0.0510 (14)
H2A	1.0258	0.4041	1.4175	0.061*
N3	1.4809 (3)	0.3773 (4)	1.5032 (3)	0.0499 (14)
H3A	1.4274	0.3748	1.4883	0.060*
N4	1.2184 (4)	0.4345 (5)	1.4465 (3)	0.0754 (18)*
N5	1.1129 (5)	0.3612 (5)	1.3056 (4)	0.094 (2)*
N6	1.2775 (4)	0.2273 (5)	1.5051 (3)	0.0677 (17)*
S1	1.25218 (13)	0.60404 (18)	1.51067 (13)	0.0996 (9)
S2	0.94976 (19)	0.3689 (2)	1.23046 (17)	0.1467 (14)
S3	1.3324 (2)	0.1029 (2)	1.60350 (12)	0.1132 (10)
O1	1.1632 (3)	0.1444 (4)	1.3004 (2)	0.0752 (15)
O2	1.3015 (3)	0.1459 (3)	1.3889 (2)	0.0561 (12)
O3	1.1243 (3)	0.1167 (3)	1.4178 (2)	0.0674 (14)
O4	1.0848 (3)	0.2913 (3)	1.4208 (2)	0.0595 (13)
O5	1.2824 (3)	0.3020 (4)	1.2938 (2)	0.0641 (13)
O6	1.3500 (2)	0.3412 (3)	1.4114 (2)	0.0538 (12)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Nd	0.0425 (2)	0.0534 (3)	0.0489 (3)	0.00325 (17)	0.00670 (16)	0.0035 (2)
C1	0.084 (7)	0.122 (9)	0.152 (11)	0.005 (6)	-0.028 (7)	0.001 (8)
C2	0.073 (6)	0.069 (6)	0.104 (8)	0.013 (5)	-0.016 (6)	-0.005 (6)
C3	0.053 (5)	0.092 (7)	0.134 (10)	0.022 (5)	0.000 (6)	0.007 (7)
C4	0.066 (5)	0.084 (6)	0.088 (7)	0.018 (4)	0.019 (5)	0.007 (5)
C5	0.072 (5)	0.040 (4)	0.078 (6)	0.018 (4)	0.016 (5)	0.015 (4)
C6	0.061 (5)	0.068 (5)	0.063 (5)	0.001 (4)	0.009 (4)	0.007 (4)
C7	0.091 (6)	0.059 (5)	0.083 (7)	0.012 (5)	0.001 (5)	0.010 (5)
C8	0.074 (5)	0.048 (5)	0.073 (6)	0.015 (4)	0.023 (5)	0.001 (4)
C9	0.092 (6)	0.062 (5)	0.058 (5)	0.003 (4)	0.029 (5)	-0.010 (4)
C10	0.116 (8)	0.086 (7)	0.099 (8)	0.001 (6)	0.045 (7)	-0.031 (6)
C11	0.140 (10)	0.117 (9)	0.074 (7)	-0.021 (8)	0.031 (7)	-0.051 (6)
C12	0.118 (8)	0.094 (7)	0.073 (7)	-0.016 (6)	0.023 (6)	-0.016 (6)
C13	0.084 (6)	0.059 (5)	0.050 (5)	-0.002 (4)	0.013 (4)	-0.005 (4)

supplementary materials

C14	0.076 (5)	0.050 (4)	0.057 (5)	0.002 (4)	0.018 (4)	0.004 (4)
C15	0.079 (6)	0.099 (7)	0.086 (7)	-0.020 (5)	-0.029 (5)	0.007 (6)
C16	0.122 (7)	0.054 (5)	0.085 (7)	0.000 (5)	0.003 (6)	0.002 (5)
C17	0.078 (5)	0.047 (5)	0.049 (5)	0.005 (4)	-0.003 (4)	-0.002 (4)
C18	0.074 (5)	0.065 (5)	0.067 (5)	0.011 (4)	0.019 (4)	-0.001 (4)
C19	0.062 (4)	0.055 (5)	0.065 (5)	-0.003 (3)	0.021 (4)	0.002 (4)
C20	0.055 (4)	0.052 (4)	0.049 (4)	-0.006 (3)	0.012 (3)	-0.001 (3)
C21	0.056 (4)	0.054 (5)	0.066 (5)	0.001 (4)	0.009 (4)	0.000 (4)
C22	0.063 (5)	0.057 (5)	0.083 (6)	-0.006 (4)	0.003 (4)	0.011 (5)
C23	0.052 (4)	0.060 (5)	0.075 (5)	-0.001 (4)	0.023 (4)	-0.006 (4)
C24	0.053 (4)	0.063 (5)	0.059 (5)	-0.004 (3)	0.018 (4)	-0.002 (4)
C25	0.076 (5)	0.072 (6)	0.136 (9)	-0.014 (5)	0.056 (6)	-0.012 (6)
C26	0.091 (7)	0.070 (7)	0.161 (10)	-0.026 (5)	0.046 (7)	-0.023 (7)
C27	0.096 (6)	0.047 (5)	0.104 (7)	-0.007 (5)	0.027 (6)	-0.002 (5)
C28	0.063 (4)	0.036 (4)	0.079 (6)	-0.004 (3)	0.017 (4)	0.003 (4)
C29	0.056 (4)	0.051 (4)	0.042 (4)	-0.012 (3)	0.013 (3)	0.003 (3)
C30	0.098 (6)	0.042 (5)	0.128 (8)	0.020 (4)	0.045 (6)	0.014 (5)
C31	0.189 (11)	0.114 (9)	0.051 (6)	-0.040 (8)	0.012 (7)	-0.017 (6)
C32	0.111 (7)	0.070 (6)	0.049 (5)	-0.015 (5)	0.013 (5)	-0.015 (4)
C33	0.112 (7)	0.090 (7)	0.064 (6)	-0.031 (5)	0.042 (6)	-0.018 (5)
C34	0.074 (5)	0.088 (6)	0.062 (6)	-0.017 (4)	0.026 (5)	-0.017 (5)
C35	0.057 (4)	0.046 (4)	0.050 (4)	-0.004 (3)	0.011 (4)	-0.007 (3)
C36	0.057 (4)	0.072 (5)	0.060 (5)	-0.013 (4)	0.018 (4)	-0.006 (4)
C37	0.075 (5)	0.077 (6)	0.063 (6)	-0.010 (4)	0.001 (5)	-0.022 (5)
C38	0.048 (4)	0.053 (4)	0.058 (5)	0.003 (3)	0.018 (4)	-0.004 (4)
C39	0.050 (4)	0.045 (4)	0.050 (4)	-0.002 (3)	0.015 (3)	0.003 (3)
C40	0.052 (4)	0.074 (5)	0.071 (6)	-0.003 (4)	0.024 (4)	-0.016 (4)
C41	0.077 (5)	0.088 (6)	0.061 (6)	-0.002 (4)	0.034 (5)	-0.011 (5)
C42	0.074 (5)	0.065 (5)	0.042 (4)	-0.003 (4)	0.013 (4)	0.005 (4)
C43	0.060 (4)	0.054 (4)	0.040 (4)	-0.007 (3)	0.009 (3)	0.002 (3)
C44	0.053 (4)	0.046 (4)	0.046 (4)	-0.002 (3)	0.013 (3)	0.006 (3)
C45	0.078 (6)	0.133 (8)	0.054 (5)	-0.017 (5)	0.001 (4)	0.025 (5)
C46	0.045 (4)	0.064 (5)	0.066 (5)	-0.010 (3)	0.012 (4)	-0.007 (4)
C47	0.079 (6)	0.102 (7)	0.080 (7)	0.013 (5)	0.007 (5)	-0.003 (5)
C48	0.068 (4)	0.064 (5)	0.043 (4)	0.012 (4)	0.018 (4)	-0.003 (4)
N1	0.062 (4)	0.050 (4)	0.074 (5)	0.010 (3)	0.016 (4)	0.003 (4)
N2	0.044 (3)	0.047 (3)	0.062 (4)	0.005 (3)	0.014 (3)	0.003 (3)
N3	0.044 (3)	0.052 (3)	0.052 (4)	0.000 (2)	0.010 (3)	-0.006 (3)
S1	0.0686 (13)	0.0964 (17)	0.145 (2)	-0.0332 (12)	0.0477 (15)	-0.0558 (17)
S2	0.094 (2)	0.142 (3)	0.161 (3)	0.0396 (18)	-0.038 (2)	-0.016 (2)
S3	0.158 (3)	0.122 (2)	0.0708 (17)	0.066 (2)	0.0515 (17)	0.0365 (16)
O1	0.073 (3)	0.082 (4)	0.059 (4)	-0.009 (3)	-0.002 (3)	0.003 (3)
O2	0.059 (3)	0.061 (3)	0.044 (3)	0.010 (2)	0.007 (2)	-0.003 (2)
O3	0.066 (3)	0.048 (3)	0.089 (4)	0.009 (2)	0.021 (3)	0.005 (3)
O4	0.051 (3)	0.046 (3)	0.084 (4)	-0.001 (2)	0.023 (3)	-0.002 (3)
O5	0.060 (3)	0.084 (4)	0.043 (3)	-0.012 (2)	0.005 (2)	0.006 (3)
O6	0.048 (2)	0.069 (3)	0.046 (3)	-0.002 (2)	0.015 (2)	-0.009 (2)

Geometric parameters (Å, °)

Nd—O4	2.375 (4)	C23—H23A	0.9300
Nd—O6	2.388 (4)	C24—C25	1.421 (10)
Nd—O2	2.393 (4)	C24—C29	1.423 (9)
Nd—N5	2.489 (8)	C25—C26	1.336 (12)
Nd—N6	2.517 (7)	C25—H25A	0.9300
Nd—N4	2.530 (7)	C26—C27	1.385 (12)
Nd—O1	2.742 (5)	C26—H26A	0.9300
Nd—O5	2.778 (5)	C27—C28	1.359 (10)
Nd—O3	2.843 (5)	C27—H27A	0.9300
C1—C2	1.530 (12)	C28—O3	1.357 (8)
C1—H1B	0.9600	C28—C29	1.400 (9)
C1—H1C	0.9600	C29—O4	1.302 (7)
C1—H1D	0.9600	C30—O3	1.417 (8)
C2—C3	1.366 (12)	C30—H30A	0.9600
C2—C7	1.397 (12)	C30—H30B	0.9600
C3—C4	1.376 (12)	C30—H30C	0.9600
C3—H3B	0.9300	C31—C32	1.518 (10)
C4—C5	1.380 (10)	C31—H31A	0.9600
C4—H4A	0.9300	C31—H31B	0.9600
C5—C6	1.379 (10)	C31—H31C	0.9600
C5—N1	1.405 (9)	C32—C37	1.369 (10)
C6—C7	1.363 (10)	C32—C33	1.380 (11)
C6—H6A	0.9300	C33—C34	1.379 (10)
C7—H7A	0.9300	C33—H33A	0.9300
C8—N1	1.289 (9)	C34—C35	1.350 (9)
C8—C9	1.413 (10)	C34—H34A	0.9300
C8—H8A	0.9300	C35—C36	1.379 (9)
C9—C14	1.403 (10)	C35—N3	1.427 (8)
C9—C10	1.435 (11)	C36—C37	1.388 (9)
C10—C11	1.358 (13)	C36—H36A	0.9300
C10—H10A	0.9300	C37—H37A	0.9300
C11—C12	1.380 (13)	C38—N3	1.311 (7)
C11—H11A	0.9300	C38—C39	1.400 (9)
C12—C13	1.373 (11)	C38—H38A	0.9300
C12—H12A	0.9300	C39—C40	1.409 (9)
C13—O1	1.358 (9)	C39—C44	1.410 (8)
C13—C14	1.412 (10)	C40—C41	1.347 (10)
C14—O2	1.307 (8)	C40—H40A	0.9300
C15—O1	1.450 (8)	C41—C42	1.403 (10)
C15—H15A	0.9600	C41—H41A	0.9300
C15—H15B	0.9600	C42—C43	1.369 (9)
C15—H15C	0.9600	C42—H42A	0.9300
C16—C17	1.503 (10)	C43—O5	1.371 (8)
C16—H16A	0.9600	C43—C44	1.403 (9)
C16—H16B	0.9600	C44—O6	1.301 (7)
C16—H16C	0.9600	C45—O5	1.433 (8)

supplementary materials

C17—C18	1.374 (10)	C45—H45A	0.9600
C17—C22	1.380 (10)	C45—H45B	0.9600
C18—C19	1.391 (10)	C45—H45C	0.9600
C18—H18A	0.9300	C46—N4	1.150 (8)
C19—C20	1.355 (9)	C46—S1	1.630 (8)
C19—H19A	0.9300	C47—N5	1.171 (10)
C20—C21	1.379 (9)	C47—S2	1.594 (9)
C20—N2	1.426 (8)	C48—N6	1.145 (8)
C21—C22	1.379 (10)	C48—S3	1.617 (8)
C21—H21A	0.9300	N1—H1A	0.8600
C22—H22A	0.9300	N2—H2A	0.8600
C23—N2	1.284 (8)	N3—H3A	0.8600
C23—C24	1.408 (9)		
O4—Nd—O6	143.38 (15)	C21—C22—C17	122.1 (7)
O4—Nd—O2	133.12 (15)	C21—C22—H22A	119.0
O6—Nd—O2	74.67 (15)	C17—C22—H22A	119.0
O4—Nd—N5	73.0 (2)	N2—C23—C24	123.4 (6)
O6—Nd—N5	110.2 (2)	N2—C23—H23A	118.3
O2—Nd—N5	129.1 (2)	C24—C23—H23A	118.3
O4—Nd—N6	86.67 (18)	C23—C24—C25	120.9 (7)
O6—Nd—N6	79.11 (18)	C23—C24—C29	120.9 (6)
O2—Nd—N6	73.77 (18)	C25—C24—C29	118.2 (7)
N5—Nd—N6	156.4 (2)	C26—C25—C24	120.3 (8)
O4—Nd—N4	73.90 (18)	C26—C25—H25A	119.8
O6—Nd—N4	70.59 (17)	C24—C25—H25A	119.8
O2—Nd—N4	139.96 (18)	C25—C26—C27	120.6 (8)
N5—Nd—N4	82.2 (2)	C25—C26—H26A	119.7
N6—Nd—N4	80.6 (2)	C27—C26—H26A	119.7
O4—Nd—O1	98.57 (16)	C28—C27—C26	122.4 (8)
O6—Nd—O1	117.78 (15)	C28—C27—H27A	118.8
O2—Nd—O1	59.79 (16)	C26—C27—H27A	118.8
N5—Nd—O1	75.5 (2)	O3—C28—C27	128.3 (7)
N6—Nd—O1	120.39 (18)	O3—C28—C29	113.2 (6)
N4—Nd—O1	157.76 (19)	C27—C28—C29	118.5 (7)
O4—Nd—O5	142.12 (16)	O4—C29—C28	119.4 (6)
O6—Nd—O5	59.78 (14)	O4—C29—C24	120.7 (6)
O2—Nd—O5	70.97 (15)	C28—C29—C24	119.9 (6)
N5—Nd—O5	69.7 (2)	O3—C30—H30A	109.5
N6—Nd—O5	131.17 (17)	O3—C30—H30B	109.5
N4—Nd—O5	106.50 (18)	H30A—C30—H30B	109.5
O1—Nd—O5	66.39 (15)	O3—C30—H30C	109.5
O4—Nd—O3	58.38 (13)	H30A—C30—H30C	109.5
O6—Nd—O3	142.73 (15)	H30B—C30—H30C	109.5
O2—Nd—O3	75.00 (15)	C32—C31—H31A	109.5
N5—Nd—O3	105.6 (2)	C32—C31—H31B	109.5
N6—Nd—O3	72.00 (18)	H31A—C31—H31B	109.5
N4—Nd—O3	125.27 (17)	C32—C31—H31C	109.5
O1—Nd—O3	61.95 (15)	H31A—C31—H31C	109.5
O5—Nd—O3	127.32 (14)	H31B—C31—H31C	109.5

C2—C1—H1B	109.5	C37—C32—C33	118.4 (8)
C2—C1—H1C	109.5	C37—C32—C31	119.6 (8)
H1B—C1—H1C	109.5	C33—C32—C31	122.0 (9)
C2—C1—H1D	109.5	C34—C33—C32	121.2 (8)
H1B—C1—H1D	109.5	C34—C33—H33A	119.4
H1C—C1—H1D	109.5	C32—C33—H33A	119.4
C3—C2—C7	117.3 (9)	C35—C34—C33	119.8 (8)
C3—C2—C1	122.1 (10)	C35—C34—H34A	120.1
C7—C2—C1	120.7 (11)	C33—C34—H34A	120.1
C2—C3—C4	122.3 (9)	C34—C35—C36	120.6 (7)
C2—C3—H3B	118.8	C34—C35—N3	118.6 (6)
C4—C3—H3B	118.8	C36—C35—N3	120.8 (6)
C3—C4—C5	119.2 (9)	C35—C36—C37	119.3 (7)
C3—C4—H4A	120.4	C35—C36—H36A	120.3
C5—C4—H4A	120.4	C37—C36—H36A	120.3
C6—C5—C4	119.8 (8)	C32—C37—C36	120.8 (7)
C6—C5—N1	117.6 (7)	C32—C37—H37A	119.6
C4—C5—N1	122.5 (8)	C36—C37—H37A	119.6
C7—C6—C5	119.8 (8)	N3—C38—C39	123.9 (6)
C7—C6—H6A	120.1	N3—C38—H38A	118.0
C5—C6—H6A	120.1	C39—C38—H38A	118.0
C6—C7—C2	121.6 (9)	C38—C39—C40	119.9 (6)
C6—C7—H7A	119.2	C38—C39—C44	120.4 (6)
C2—C7—H7A	119.2	C40—C39—C44	119.7 (6)
N1—C8—C9	122.6 (7)	C41—C40—C39	121.1 (7)
N1—C8—H8A	118.7	C41—C40—H40A	119.5
C9—C8—H8A	118.7	C39—C40—H40A	119.5
C14—C9—C8	120.7 (7)	C40—C41—C42	120.3 (7)
C14—C9—C10	118.0 (8)	C40—C41—H41A	119.9
C8—C9—C10	121.3 (8)	C42—C41—H41A	119.9
C11—C10—C9	120.7 (9)	C43—C42—C41	119.4 (7)
C11—C10—H10A	119.6	C43—C42—H42A	120.3
C9—C10—H10A	119.6	C41—C42—H42A	120.3
C10—C11—C12	120.7 (9)	C42—C43—O5	125.0 (6)
C10—C11—H11A	119.7	C42—C43—C44	122.2 (6)
C12—C11—H11A	119.7	O5—C43—C44	112.8 (6)
C13—C12—C11	120.8 (10)	O6—C44—C43	120.4 (6)
C13—C12—H12A	119.6	O6—C44—C39	122.2 (6)
C11—C12—H12A	119.6	C43—C44—C39	117.4 (6)
O1—C13—C12	126.4 (8)	O5—C45—H45A	109.5
O1—C13—C14	113.4 (7)	O5—C45—H45B	109.5
C12—C13—C14	120.1 (8)	H45A—C45—H45B	109.5
O2—C14—C9	122.0 (7)	O5—C45—H45C	109.5
O2—C14—C13	118.4 (7)	H45A—C45—H45C	109.5
C9—C14—C13	119.6 (7)	H45B—C45—H45C	109.5
O1—C15—H15A	109.5	N4—C46—S1	177.7 (7)
O1—C15—H15B	109.5	N5—C47—S2	179.4 (11)
H15A—C15—H15B	109.5	N6—C48—S3	179.1 (7)
O1—C15—H15C	109.5	C8—N1—C5	128.6 (7)

supplementary materials

H15A—C15—H15C	109.5	C8—N1—H1A	115.7
H15B—C15—H15C	109.5	C5—N1—H1A	115.7
C17—C16—H16A	109.5	C23—N2—C20	128.4 (6)
C17—C16—H16B	109.5	C23—N2—H2A	115.8
H16A—C16—H16B	109.5	C20—N2—H2A	115.8
C17—C16—H16C	109.5	C38—N3—C35	128.9 (6)
H16A—C16—H16C	109.5	C38—N3—H3A	115.6
H16B—C16—H16C	109.5	C35—N3—H3A	115.6
C18—C17—C22	117.5 (7)	C46—N4—Nd	169.5 (6)
C18—C17—C16	121.3 (7)	C47—N5—Nd	145.5 (7)
C22—C17—C16	121.3 (7)	C48—N6—Nd	156.7 (6)
C17—C18—C19	121.4 (7)	C13—O1—C15	118.2 (7)
C17—C18—H18A	119.3	C13—O1—Nd	115.5 (4)
C19—C18—H18A	119.3	C15—O1—Nd	126.1 (5)
C20—C19—C18	119.5 (7)	C14—O2—Nd	127.0 (4)
C20—C19—H19A	120.2	C28—O3—C30	118.4 (6)
C18—C19—H19A	120.2	C28—O3—Nd	114.2 (4)
C19—C20—C21	120.8 (7)	C30—O3—Nd	127.1 (4)
C19—C20—N2	121.5 (6)	C29—O4—Nd	130.7 (4)
C21—C20—N2	117.6 (6)	C43—O5—C45	117.5 (6)
C20—C21—C22	118.7 (7)	C43—O5—Nd	113.7 (4)
C20—C21—H21A	120.6	C45—O5—Nd	128.6 (4)
C22—C21—H21A	120.6	C44—O6—Nd	126.7 (4)
C7—C2—C3—C4	0.6 (14)	O5—Nd—N5—C47	131.5 (13)
C1—C2—C3—C4	-178.2 (9)	O3—Nd—N5—C47	6.9 (13)
C2—C3—C4—C5	-0.5 (14)	O4—Nd—N6—C48	93.5 (14)
C3—C4—C5—C6	0.1 (12)	O6—Nd—N6—C48	-120.4 (14)
C3—C4—C5—N1	177.5 (7)	O2—Nd—N6—C48	-43.4 (14)
C4—C5—C6—C7	0.0 (11)	N5—Nd—N6—C48	123.6 (14)
N1—C5—C6—C7	-177.5 (7)	N4—Nd—N6—C48	167.7 (14)
C5—C6—C7—C2	0.2 (12)	O1—Nd—N6—C48	-4.5 (15)
C3—C2—C7—C6	-0.5 (13)	O5—Nd—N6—C48	-88.5 (14)
C1—C2—C7—C6	178.4 (8)	O3—Nd—N6—C48	35.7 (14)
N1—C8—C9—C14	0.5 (12)	C12—C13—O1—C15	11.3 (12)
N1—C8—C9—C10	179.5 (8)	C14—C13—O1—C15	-170.5 (7)
C14—C9—C10—C11	0.1 (13)	C12—C13—O1—Nd	-163.8 (7)
C8—C9—C10—C11	-178.9 (9)	C14—C13—O1—Nd	14.4 (8)
C9—C10—C11—C12	0.8 (16)	O4—Nd—O1—C13	-153.4 (5)
C10—C11—C12—C13	-0.5 (16)	O6—Nd—O1—C13	31.3 (5)
C11—C12—C13—O1	177.3 (9)	O2—Nd—O1—C13	-18.0 (4)
C11—C12—C13—C14	-0.8 (13)	N5—Nd—O1—C13	136.8 (5)
C8—C9—C14—O2	-2.0 (11)	N6—Nd—O1—C13	-62.2 (5)
C10—C9—C14—O2	179.0 (7)	N4—Nd—O1—C13	138.5 (6)
C8—C9—C14—C13	177.7 (7)	O5—Nd—O1—C13	63.0 (5)
C10—C9—C14—C13	-1.3 (11)	O3—Nd—O1—C13	-106.3 (5)
O1—C13—C14—O2	3.0 (10)	O4—Nd—O1—C15	32.0 (6)
C12—C13—C14—O2	-178.7 (7)	O6—Nd—O1—C15	-143.3 (6)
O1—C13—C14—C9	-176.6 (7)	O2—Nd—O1—C15	167.4 (6)
C12—C13—C14—C9	1.7 (11)	N5—Nd—O1—C15	-37.8 (6)

C22—C17—C18—C19	-0.4 (12)	N6—Nd—O1—C15	123.2 (6)
C16—C17—C18—C19	-179.8 (7)	N4—Nd—O1—C15	-36.1 (8)
C17—C18—C19—C20	-0.3 (12)	O5—Nd—O1—C15	-111.6 (6)
C18—C19—C20—C21	1.8 (11)	O3—Nd—O1—C15	79.1 (6)
C18—C19—C20—N2	178.8 (6)	C9—C14—O2—Nd	155.3 (5)
C19—C20—C21—C22	-2.5 (11)	C13—C14—O2—Nd	-24.4 (9)
N2—C20—C21—C22	-179.6 (6)	O4—Nd—O2—C14	94.2 (6)
C20—C21—C22—C17	1.8 (12)	O6—Nd—O2—C14	-113.8 (5)
C18—C17—C22—C21	-0.4 (12)	N5—Nd—O2—C14	-10.0 (6)
C16—C17—C22—C21	179.0 (7)	N6—Nd—O2—C14	163.3 (6)
N2—C23—C24—C25	-177.4 (8)	N4—Nd—O2—C14	-144.3 (5)
N2—C23—C24—C29	3.1 (12)	O1—Nd—O2—C14	22.2 (5)
C23—C24—C25—C26	177.8 (9)	O5—Nd—O2—C14	-51.0 (5)
C29—C24—C25—C26	-2.7 (14)	O3—Nd—O2—C14	88.1 (5)
C24—C25—C26—C27	1.8 (16)	C27—C28—O3—C30	-9.3 (12)
C25—C26—C27—C28	0.1 (16)	C29—C28—O3—C30	170.8 (7)
C26—C27—C28—O3	179.2 (9)	C27—C28—O3—Nd	165.9 (7)
C26—C27—C28—C29	-0.9 (14)	C29—C28—O3—Nd	-14.0 (8)
O3—C28—C29—O4	0.8 (10)	O4—Nd—O3—C28	15.7 (5)
C27—C28—C29—O4	-179.1 (7)	O6—Nd—O3—C28	154.0 (4)
O3—C28—C29—C24	179.9 (6)	O2—Nd—O3—C28	-169.5 (5)
C27—C28—C29—C24	0.0 (11)	N5—Nd—O3—C28	-42.4 (5)
C23—C24—C29—O4	0.4 (11)	N6—Nd—O3—C28	113.0 (5)
C25—C24—C29—O4	-179.1 (7)	N4—Nd—O3—C28	49.1 (5)
C23—C24—C29—C28	-178.7 (7)	O1—Nd—O3—C28	-106.1 (5)
C25—C24—C29—C28	1.8 (11)	O5—Nd—O3—C28	-118.4 (5)
C37—C32—C33—C34	-1.6 (14)	O4—Nd—O3—C30	-169.6 (7)
C31—C32—C33—C34	-179.1 (9)	O6—Nd—O3—C30	-31.3 (7)
C32—C33—C34—C35	1.7 (13)	O2—Nd—O3—C30	5.2 (6)
C33—C34—C35—C36	-0.9 (12)	N5—Nd—O3—C30	132.3 (6)
C33—C34—C35—N3	176.1 (7)	N6—Nd—O3—C30	-72.3 (6)
C34—C35—C36—C37	0.0 (11)	N4—Nd—O3—C30	-136.2 (6)
N3—C35—C36—C37	-176.9 (7)	O1—Nd—O3—C30	68.6 (6)
C33—C32—C37—C36	0.8 (13)	O5—Nd—O3—C30	56.2 (6)
C31—C32—C37—C36	178.4 (8)	C28—C29—O4—Nd	18.9 (9)
C35—C36—C37—C32	0.0 (12)	C24—C29—O4—Nd	-160.2 (5)
N3—C38—C39—C40	178.8 (6)	O6—Nd—O4—C29	-155.9 (5)
N3—C38—C39—C44	-2.8 (10)	O2—Nd—O4—C29	-25.1 (7)
C38—C39—C40—C41	178.9 (7)	N5—Nd—O4—C29	103.0 (6)
C44—C39—C40—C41	0.5 (11)	N6—Nd—O4—C29	-89.1 (6)
C39—C40—C41—C42	0.7 (12)	N4—Nd—O4—C29	-170.3 (6)
C40—C41—C42—C43	-2.1 (12)	O1—Nd—O4—C29	31.1 (6)
C41—C42—C43—O5	179.5 (7)	O5—Nd—O4—C29	93.3 (6)
C41—C42—C43—C44	2.2 (11)	O3—Nd—O4—C29	-18.3 (5)
C42—C43—C44—O6	179.3 (6)	C42—C43—O5—C45	-11.0 (10)
O5—C43—C44—O6	1.7 (9)	C44—C43—O5—C45	166.5 (6)
C42—C43—C44—C39	-1.0 (10)	C42—C43—O5—Nd	163.6 (6)
O5—C43—C44—C39	-178.6 (6)	C44—C43—O5—Nd	-18.9 (7)
C38—C39—C44—O6	0.9 (10)	O4—Nd—O5—C43	160.4 (4)

supplementary materials

C40—C39—C44—O6	179.4 (6)	O6—Nd—O5—C43	20.6 (4)
C38—C39—C44—C43	-178.8 (6)	O2—Nd—O5—C43	-62.4 (4)
C40—C39—C44—C43	-0.3 (10)	N5—Nd—O5—C43	150.5 (5)
C9—C8—N1—C5	-173.3 (7)	N6—Nd—O5—C43	-16.3 (5)
C6—C5—N1—C8	-160.0 (7)	N4—Nd—O5—C43	75.6 (5)
C4—C5—N1—C8	22.5 (11)	O1—Nd—O5—C43	-126.9 (5)
C24—C23—N2—C20	178.3 (7)	O3—Nd—O5—C43	-115.0 (4)
C19—C20—N2—C23	10.6 (11)	O4—Nd—O5—C45	-25.8 (7)
C21—C20—N2—C23	-172.3 (7)	O6—Nd—O5—C45	-165.6 (7)
C39—C38—N3—C35	178.2 (6)	O2—Nd—O5—C45	111.5 (6)
C34—C35—N3—C38	-164.5 (7)	N5—Nd—O5—C45	-35.7 (6)
C36—C35—N3—C38	12.5 (10)	N6—Nd—O5—C45	157.5 (6)
O4—Nd—N4—C46	150 (3)	N4—Nd—O5—C45	-110.6 (6)
O6—Nd—N4—C46	-21 (3)	O1—Nd—O5—C45	46.9 (6)
O2—Nd—N4—C46	10 (4)	O3—Nd—O5—C45	58.8 (7)
N5—Nd—N4—C46	-136 (3)	C43—C44—O6—Nd	22.3 (9)
N6—Nd—N4—C46	61 (3)	C39—C44—O6—Nd	-157.3 (5)
O1—Nd—N4—C46	-137 (3)	O4—Nd—O6—C44	-160.7 (5)
O5—Nd—N4—C46	-70 (3)	O2—Nd—O6—C44	54.3 (5)
O3—Nd—N4—C46	121 (3)	N5—Nd—O6—C44	-72.3 (5)
O4—Nd—N5—C47	-42.1 (12)	N6—Nd—O6—C44	130.2 (5)
O6—Nd—N5—C47	176.5 (12)	N4—Nd—O6—C44	-146.0 (5)
O2—Nd—N5—C47	90.1 (13)	O1—Nd—O6—C44	11.5 (5)
N6—Nd—N5—C47	-73.7 (15)	O5—Nd—O6—C44	-22.3 (5)
N4—Nd—N5—C47	-117.6 (13)	O3—Nd—O6—C44	90.8 (5)
O1—Nd—N5—C47	61.8 (13)		

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1A \cdots O2	0.86	1.86	2.566 (7)	138
N2—H2A \cdots O4	0.86	1.87	2.576 (6)	138
N3—H3A \cdots O6	0.86	1.89	2.589 (7)	137
C38—H38A \cdots S1 ⁱ	0.93	2.72	3.633 (6)	166

Symmetry codes: (i) $-x+3, -y+1, -z+3$.

Fig. 1

